

---

# “Batch and Continuous-Flow One-Pot Processes using Amine Diazotization to Produce Silylated Diazo Reagents”

---

Audubert, C.; Gamboa Marin, O. J.; Lebel, H. *Angew. Chem. Int. Ed.* **2017**, *56*, 6294–6297.

Leila Terrab

Wipf Group

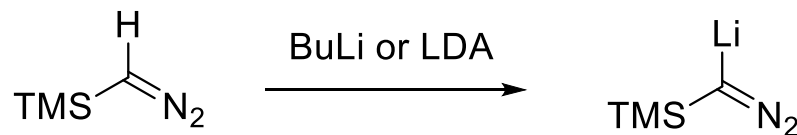
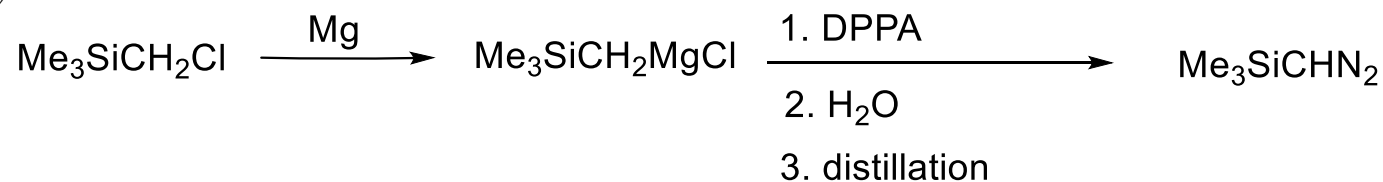
06/10/2017

# Trimethylsilyldiazomethane

*Lappert*- First report of covalent organometallic diazoalkanes:



*Shioiri and Aoyama*:



Lappert, M. F.; Lorberth, J. *Chem. Comm.* **1967**, 16, 836–837.

Shioiri, T.; Aoyama, T.; Snowden, T. *Encyclopedia of Reagents for Organic Synthesis* **2001**.

DOI:10.1002/047084289X.rt298

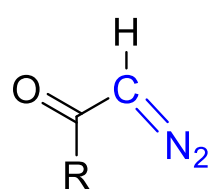
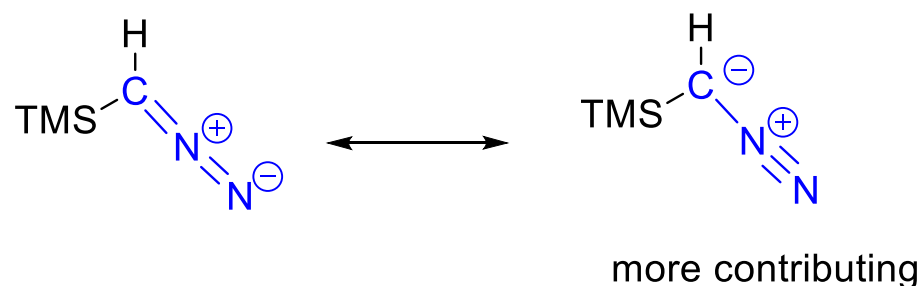
Poole, J. *J. Org. Chem.* **1998**, 340, 679.

# Arndt-Eistert Homologation

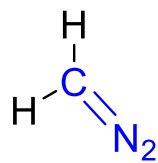
TMSCHN<sub>2</sub>, Proposed by Aoyama and Shioiri as an alternative to diazomethane

*Seyferth*:

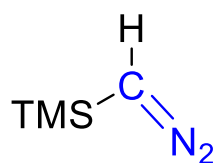
TMSCHN<sub>2</sub> is known to be thermally stable due to the C-Si p<sub>π</sub> – d<sub>π</sub> resonance:



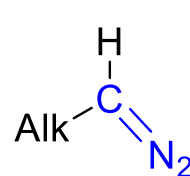
2100-2087



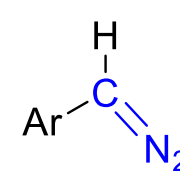
2074



2070



2049-2020



2049-2020

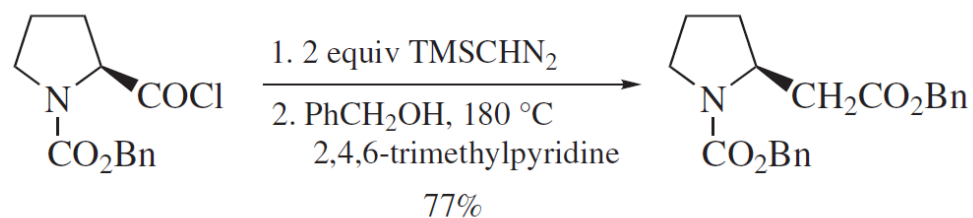
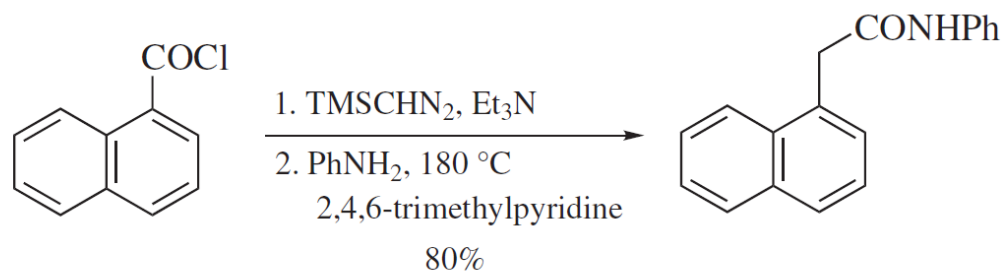
IR stretching (cm<sup>-1</sup>)

Aoyama, T.; Shioiri, T. *Tetrahedron Lett.* **1980**, *21*, 4461.

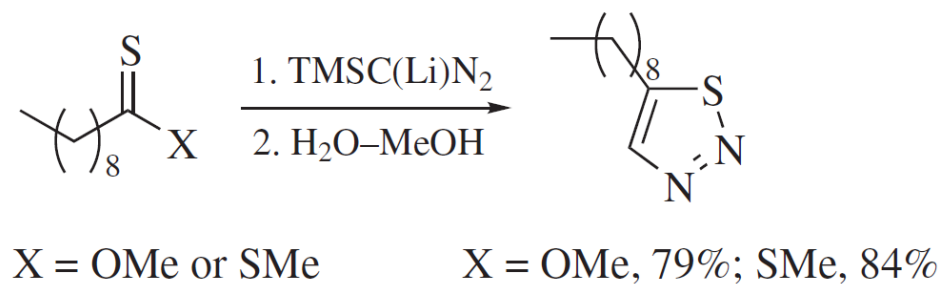
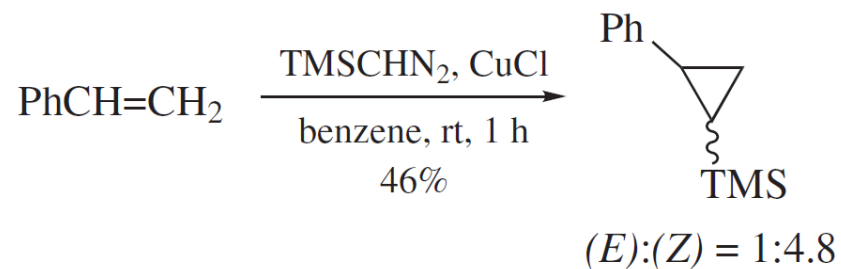
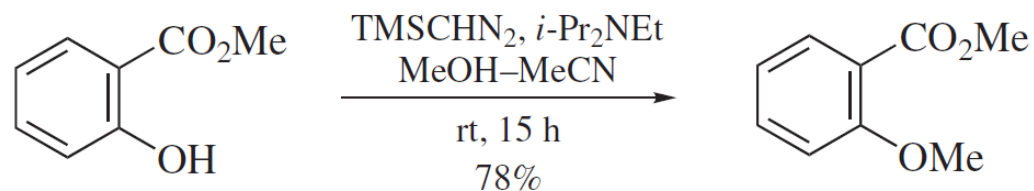
Seyferth, D.; Menzel, A.; Dow, A. W.; Flood, T. C. *J. Organomet. Chem.* **1972**, *44*, 279.

# Reactions with Trimethylsilyldiazomethane

Arndt-Eistert Homologation:

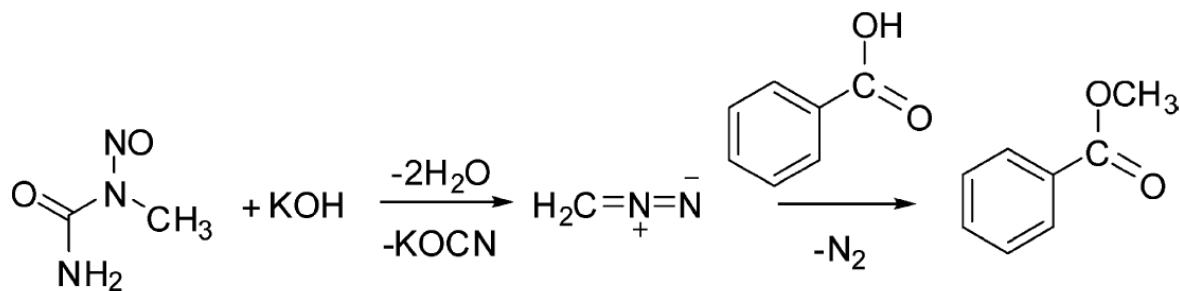
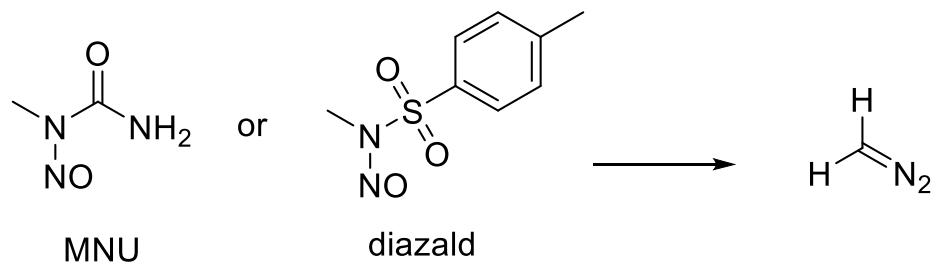


# Reactions with Trimethylsilyldiazomethane

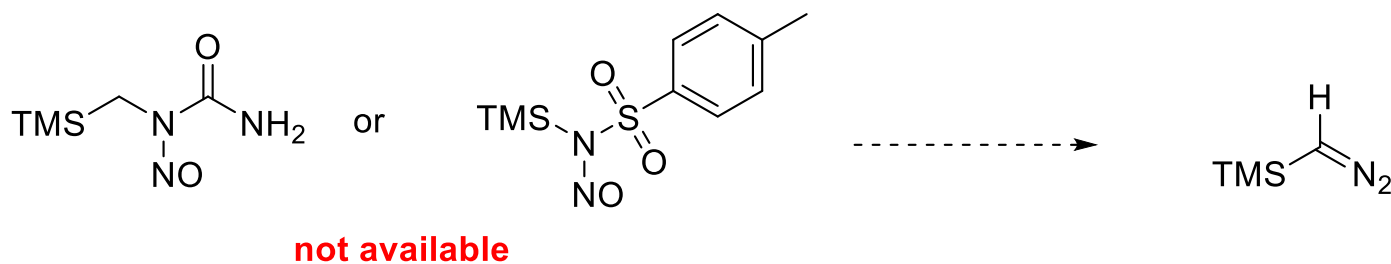


# Possible Strategies to Synthesize TMSCHN<sub>2</sub>

Previously reported, done by flow chemistry:

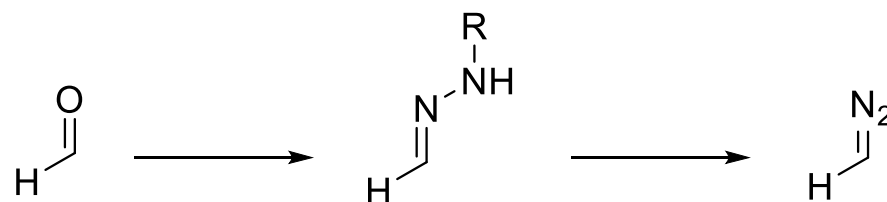


Trimethylsilylazide derivative:

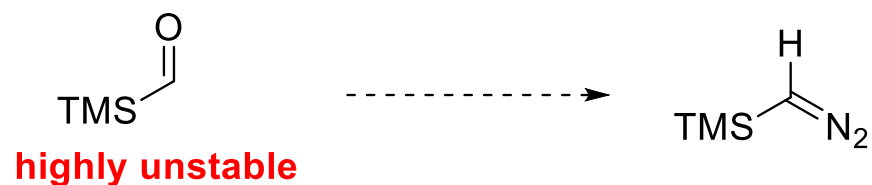


# Possible Strategies to Synthesize TMSCHN<sub>2</sub>

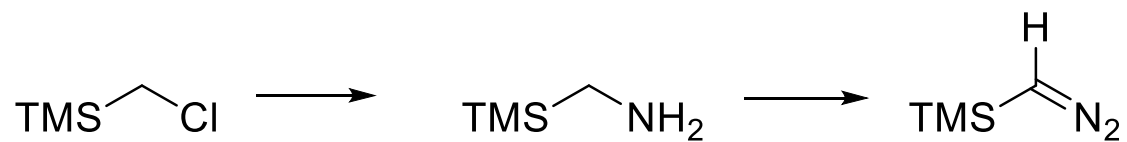
Diazomethane:



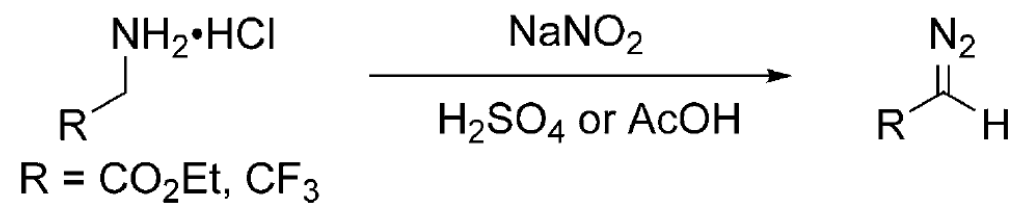
Trimethylsilylazide derivative:



Synthesis used:



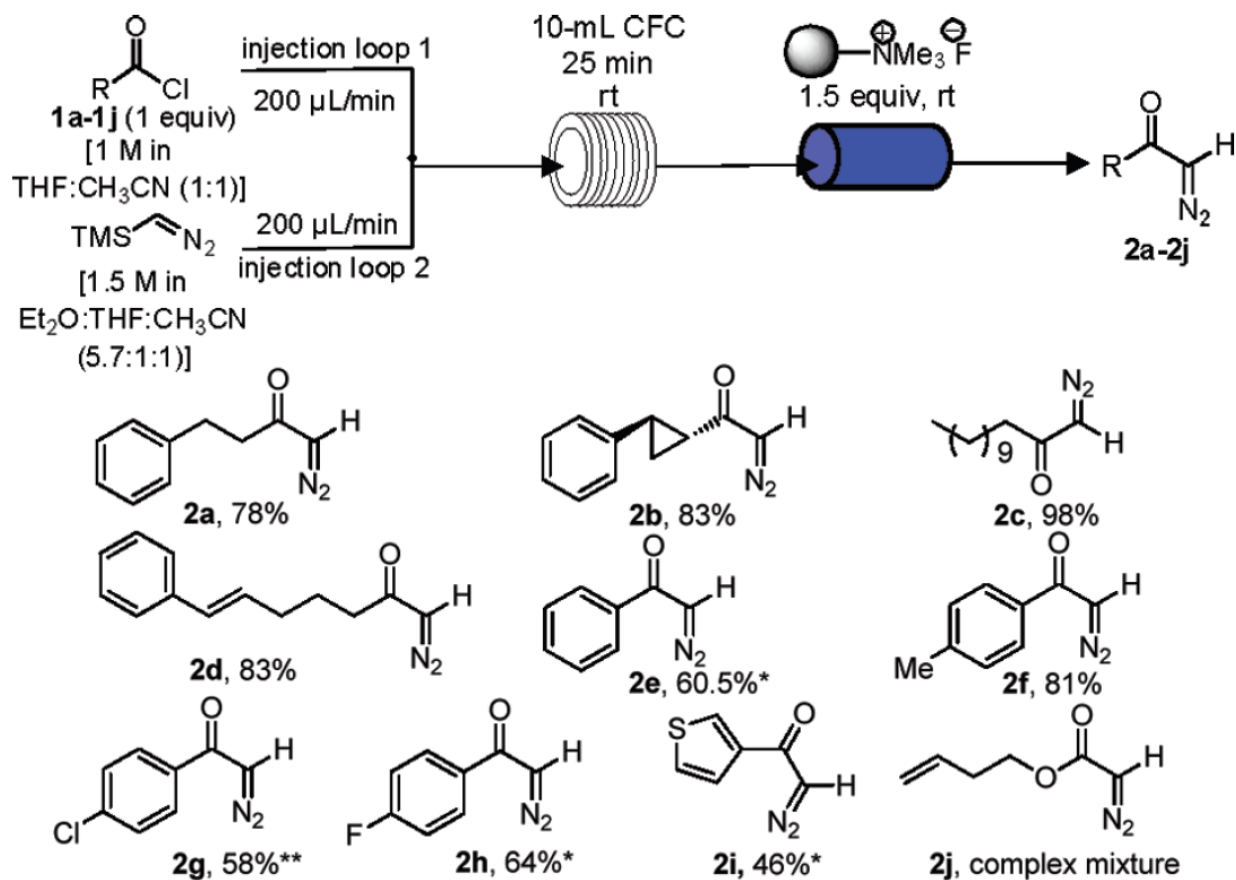
# Possible strategies to synthesize TMSCHN<sub>2</sub>



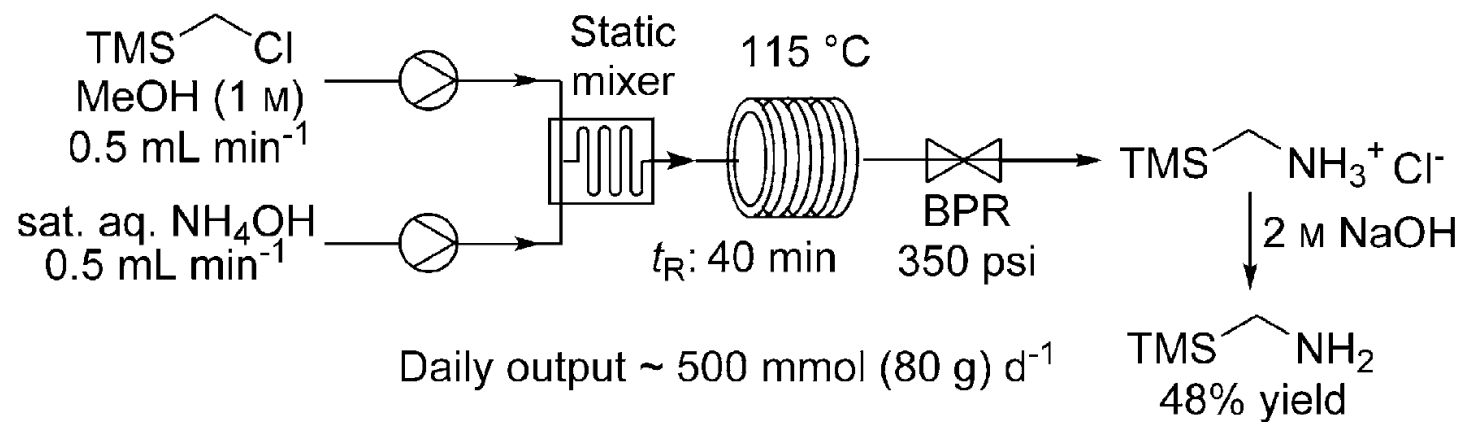


# Continuous flow of trimethylsilylazide has not been reported

**Scheme 2.** Flow Synthesis of Diazoketones<sup>a</sup>

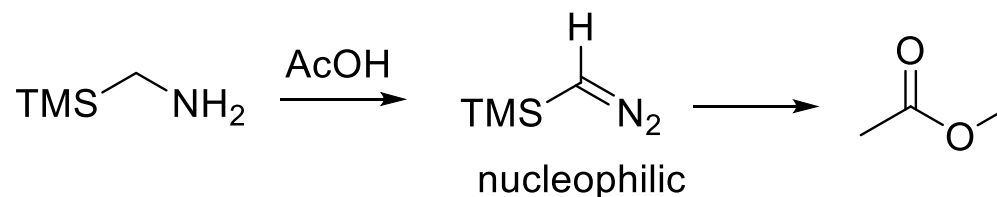


# Flow Synthesis of Intermediate

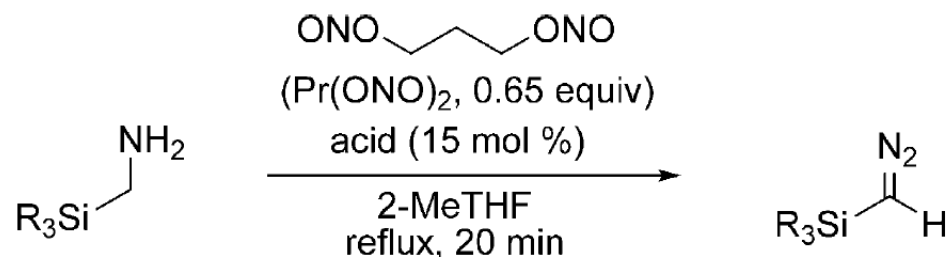


# Screening of Reaction Conditions

Acid-catalyzed diazotization



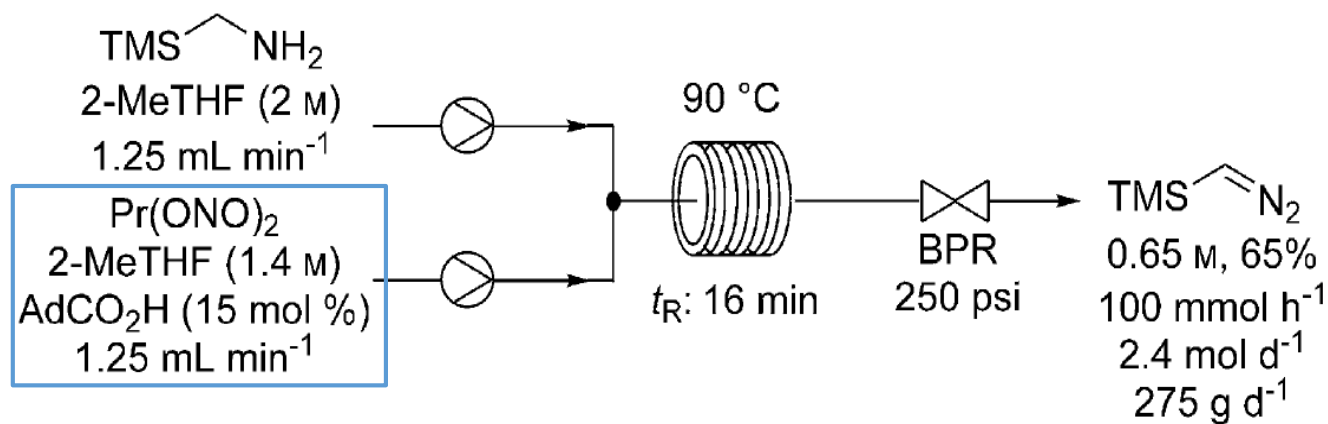
**Table 1:** Diazotization of  $\text{R}_3\text{SiCH}_2\text{NH}_2$ .



Entry	$\text{R}_3\text{Si}$	Acid	Yield [%] <sup>[a]</sup>
1	$\text{Me}_3\text{Si}$	AcOH	62
2	$\text{Me}_3\text{Si}$	AdCO <sub>2</sub> H	71
3	$\text{Me}_3\text{Si}$	4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> OH	72
4	$\text{Me}_2\text{PhSi}$	AdCO <sub>2</sub> H	63
5	$\text{Me}_2\text{PhSi}$	4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> OH	77
6	$\text{MePh}_2\text{Si}$	AdCO <sub>2</sub> H	73 <sup>[b]</sup>
7	$\text{MePh}_2\text{Si}$	4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> OH	71 <sup>[c]</sup>

[a] Yield determined by <sup>1</sup>H NMR analysis using 1,2-diphenylethane as an internal standard. [b] Reaction time = 10 min. [c] Reaction time = 40 min. THF = tetrahydrofuran.

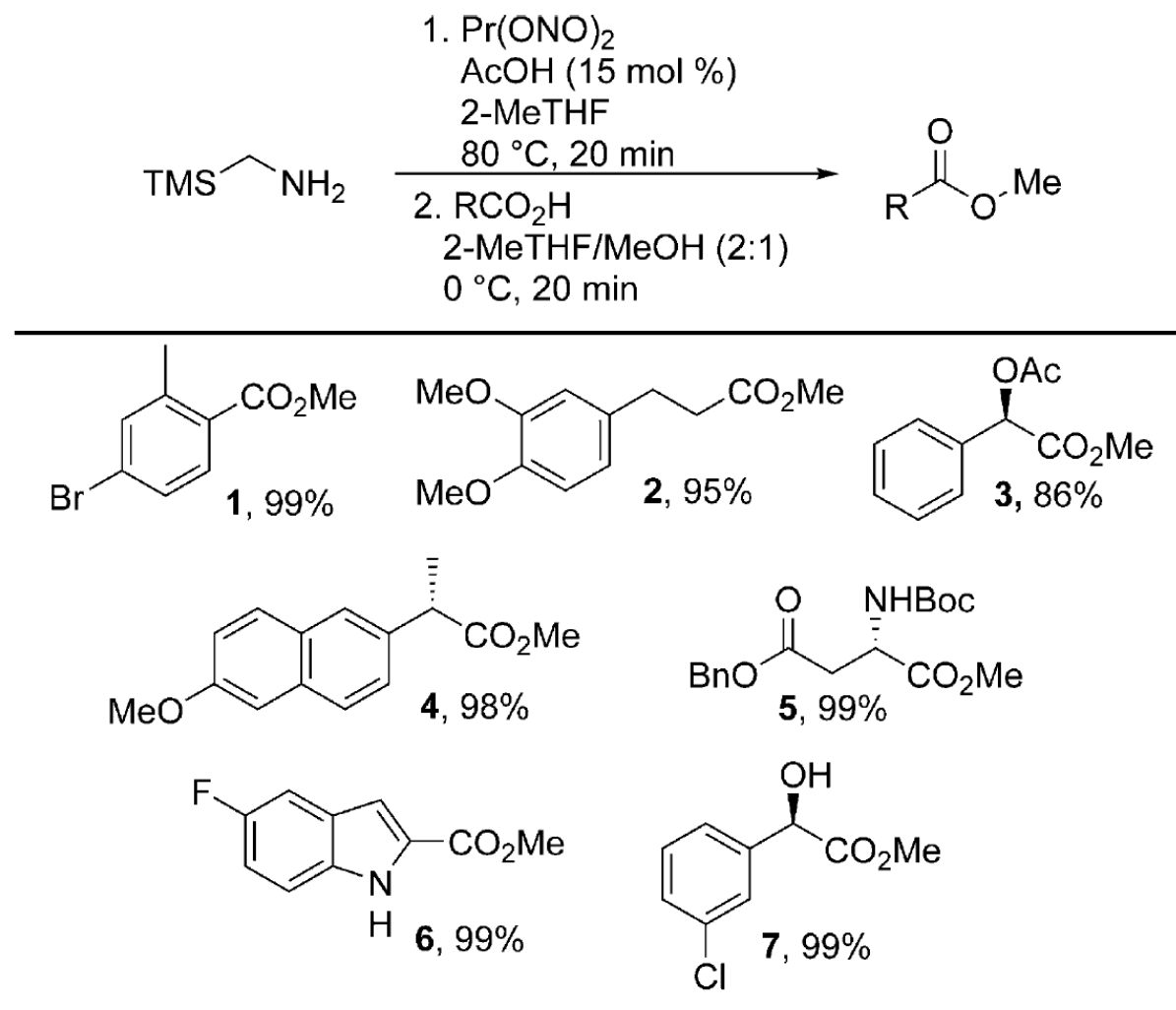
# Proposed Flow Synthesis



**Scheme 4.** Continuous-flow synthesis of TMSCH<sub>2</sub>N<sub>2</sub>.

# One-pot: Synthesis of Esters

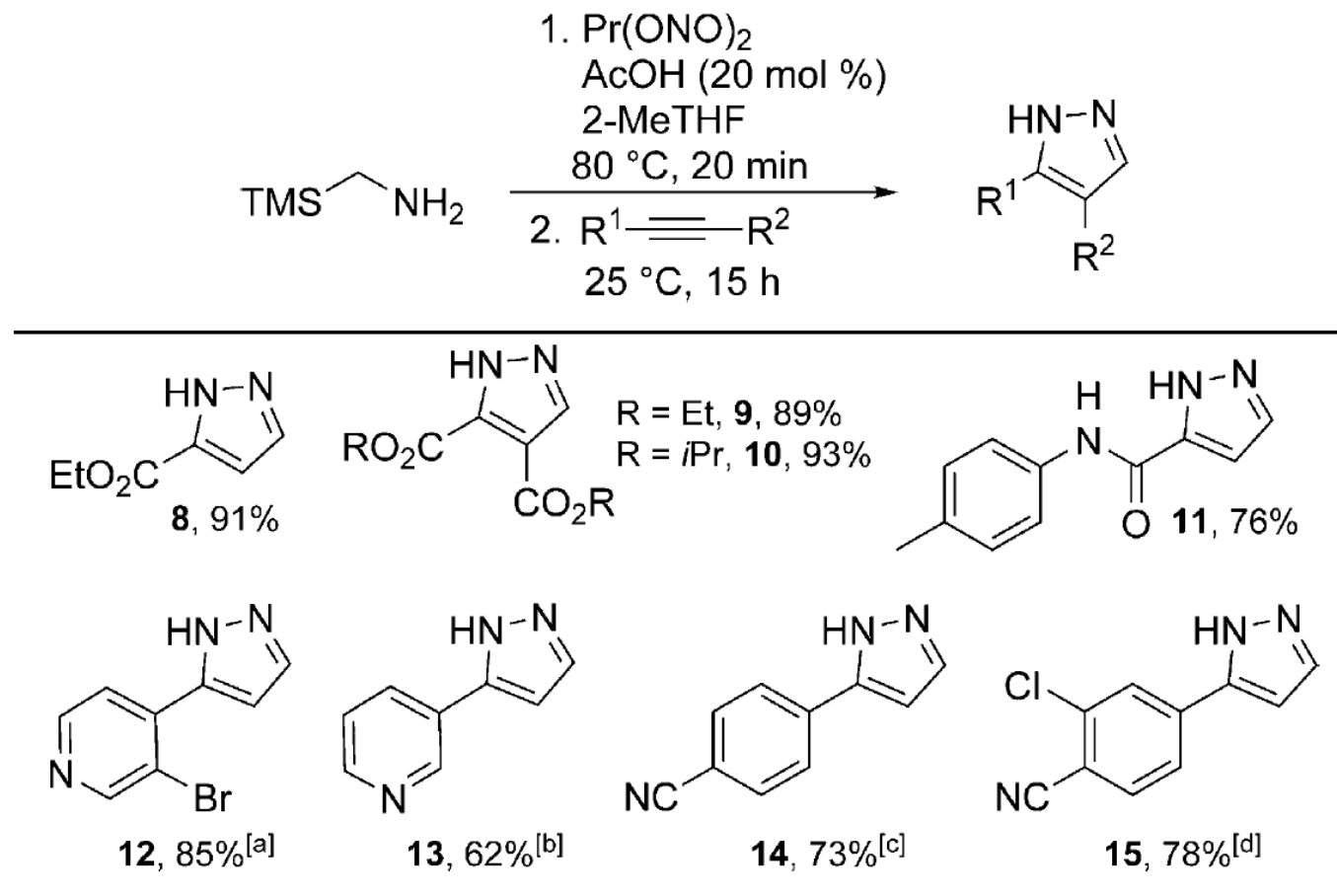
**Table 2:** One-pot esterification of carboxylic acids with TMSCH<sub>2</sub>NH<sub>2</sub>.<sup>[a]</sup>



[a] Isolated yields. Boc = *tert*-butoxycarbonyl.

# One-pot: Synthesis of Pyrazoles

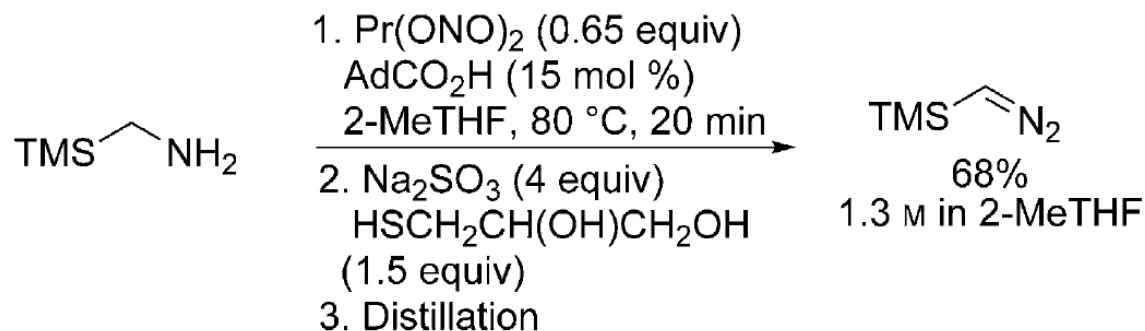
**Table 3:** One-pot 1,3-dipolar cycloaddition of alkynes with TMSCH<sub>2</sub>NH<sub>2</sub>.



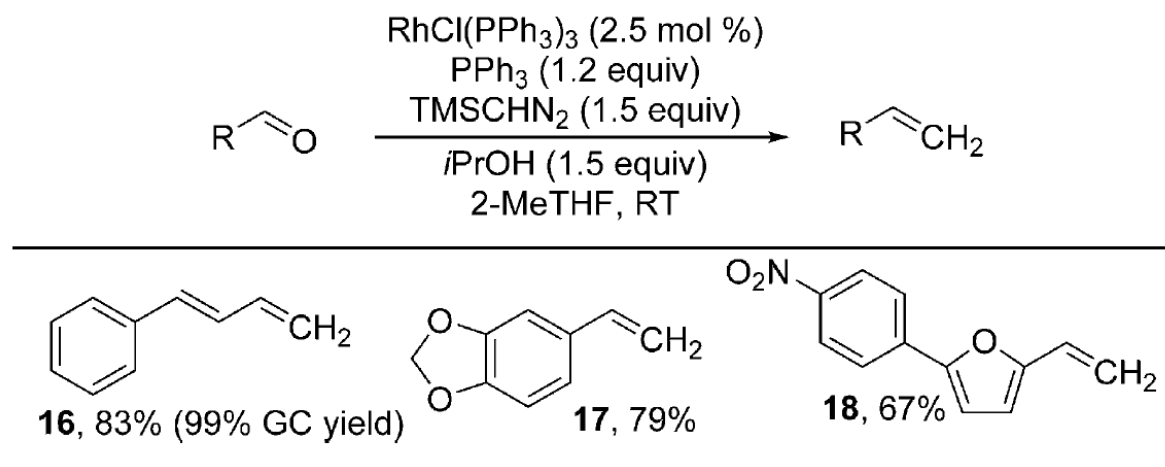
[a] RT, 60 h. [b] The 1,3-dipolar cycloaddition was performed in continuous flow at 100 °C for 30 min with a flow rate of 1.33 mL min<sup>-1</sup>. [c] 50 °C, 36 h. [d] The 1,3-dipolar cycloaddition was performed in continuous flow at 100 °C for 15 min, then 120 °C for 15 min with a flow rate of 1.33 mL min<sup>-1</sup>.

# Rhodium-catalyzed methylenation

Work-up necessary for nitrite-sensitive reagents:

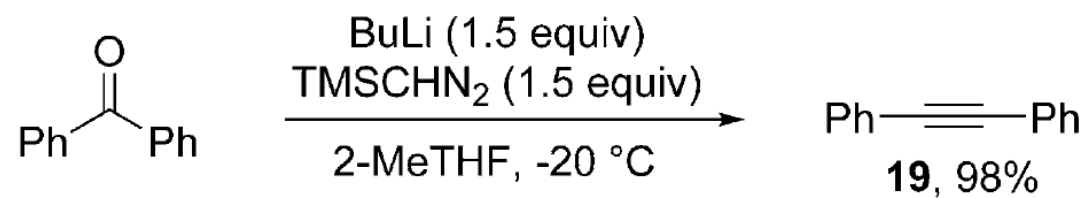


**Table 4:** Rhodium-catalyzed methylenation of aldehydes with TMSCHN<sub>2</sub> in 2-MeTHF.<sup>[a]</sup>



[a] Isolated yields.

# Homologation of ketones





# Conclusion

---

- Batch and flow synthesis of  $\text{TMSCHN}_2$  from  $\text{TMSCH}_2\text{NH}_2$  using propyldinitrite and acetic acid derivatives.
- $\text{TMSCH}_2\text{NH}_2$  used in the continuous flow reactions of carboxylic acid esterification and pyrazole formation from alkynes.
- Developed a work-up procedure of  $\text{TMSCH}_2\text{NH}_2$  to be used for Rh-catalyzed methylenation and ketone homologation.